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CHEMISTRY

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QUESTION PAPER I
M.Sc CHEMISTRY
QUESTION PAPER
2020 I

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Semester-3 ||

organometallic

transition

chemistry || **msc**

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\u0026 4 sem

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environment ,

natural product

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of spectroscopy

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Write answers as
per the given
instructions.

{ZX}e : ¶h àíZ
nÌ "A', "~' Am;a
"g' VrZ IÊS>m|
' | {d^m{OV h; &
àË¶oH\$

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VrZ IÊS>m| "A',
"~' Am;a "g' ' |
{d^m{OV h; &
àË¶oH\$ IÊS>

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SEMESTER III
PAPER XI -
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three

independent

units. The

question paper

is divided into

three parts Part-

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A, Part-B and
Part-C: Part A
(10 marks) is
compulsory and
contains 10
questions (50
words each).
Each question is
of one mark.

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SEMESTER III
PAPER XI -
SPECTROSCOPY
Page 24/102*

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Time . . . spectroscopy

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for University
of Pune Semester

4 Molecular
Structure and
Spectroscopy
question paper
gives an idea
about the
questions coming
in the board
exams and
previous years

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*Molecular
Structure and
Page 26/102*

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of one mark.

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SEMESTER III

PAPER XI -

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Time ...

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structure and sp

ectroscopy-

MSc/PhD model

question paper 1

(a) What are the

different types

of energies

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possessed by a
molecule? Why
the
translational
energy of a
molecule is not
quantized? (4
Marks) (b) Write
a note on dipole
transitions. (4
Marks)

Msc Chemistry

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Paper: ICHT-401

(Analytical
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1st May, 2018

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1.

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organic
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- I Semester -
II*

Molecular
structure and sp
ectroscopy-

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MSc/PhD model
question paper 1

(a) What are the different types of energies possessed by a molecule? Why the translational energy of a molecule is not quantized? (4 Marks) (b) Write a note on dipole

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on the Question

Paper. 2.

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Question
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ensure that they
have been
provided with
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M. Sc (Chemistry)

- 3 SEMESTER

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The major use of infrared spectroscopy is to determine the functional groups of molecules, relevant to both organic and inorganic chemistry What

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essentially a
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CHAPTER 1.

Nature of

Bonding in

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molecules:

Delocalized

Chemical

Bonding;

Conjugation;

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Cross Spectroscopy

Conjugation;
Resonance; Hyper
conjugation;
Tautomerism;
Aromaticity in
Benzenoid and
Nonbenzenoid
Compounds;
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Non-Alternant
Hydrocarbons;
Huckel's Rule:
Energy Level of

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p-Molecular
Orbitals;
Annulenes;
Antiaromaticity;
Homo-
Aromaticity; PMO
Approach; Bonds
Weaker than
Covalent;
Addition
Compounds: Crown
Ether Complexes
and Cryptands,
Inclusion

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Compounds,
Cyclodextrins;
Catenanes and
Rotaxanes

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Stereochemistry:

Chirality;

Elements of

symmetry;

Molecules with

more than one

chiral centre:

diastereomerism;

Determination of

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relative and

absolute

configuration

(octant rule

excluded) with

special

reference to

lactic acid,

alanine &

mandelic acid;

Methods of

resolution;

Optical purity;

Prochirality;

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Enantiotopic and
diastereotopic
atoms, groups
and faces;

Asymmetric
synthesis:
Cram's rule and
its

modifications,
Prelog's rule;

Conformational
analysis of
cycloalkanes
(upto six

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membered rings);
Decalins;
Conformations of
sugars; Optical
activity in
absence of
chiral carbon
(biphenyls,
allenes and
spiranes);
Chirality due to
helical shape;
Geometrical
isomerism in

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alkenes and
oximes; Methods
of determining
the

configuration

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Reaction

Mechanism:

Structure and

Reactivity:

Types of

mechanisms;

Types of

reactions;

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Thermodynamic
and kinetic
requirements;
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control;
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diagrams:
Transition
states and

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intermediates;

Methods of
determining
mechanisms;

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Hard and soft
acids and bases;

Generation,
structure,
stability and
reactivity of
carbocations,
carbanions, free
radicals,

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carbenes and
nitrenes; Effect
of structure on
reactivity; The
Hammett equation
and linear free
energy
relationship;
Substituent and
reaction
constants; Taft
equation CHAPTER
4.

Carbohydrates:

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Types of naturally occurring sugars; Deoxy sugars; Amino sugars; Branch chain sugars; General methods of determination of structure and ring size of sugars with particular reference to

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maltose,

lactose,

sucrose, starch
and cellulose.

CHAPTER 5.

Natural and

Synthetic Dyes:

Various classes

of synthetic

dyes including

heterocyclic

dyes;

Interaction

between dyes and

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fibers;
Spectroscopy

Structure
Question
elucidation of
Papers
indigo and

Alizarin CHAPTER

6. Aliphatic

Nucleophilic

Substitution:

The SN_2 , SN_1 ,

mixed SN_1 and

SN_2 , SN_i , SN_1' ,

SN_2' , SN_i' and

SET mechanisms;

The neighbouring

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mechanisms;
neighbouring
group

participation by

p and s bonds;

anchimeric

assistance;

Classical and

nonclassical

carbocations;

Phenonium ions;

Common

carbocation

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rearrangements;
Applications of
NMR spectroscopy
in the detection
of carbocations;
Reactivity-
effects of
substrate
structure,
attacking
nucleophile,
leaving group
and reaction
medium; Ambident

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nucleophiles and regioselectivity ; Phase transfer catalysis.

CHAPTER 7.

Aliphatic Electrophilic Substitution: Bimolecular mechanisms - S_N2 and S_Ni ; The S_N1 mechanism; Electrophilic substitution

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accompanied by
double bond
shifts; Effect
of substrates,
leaving group
and the solvent
polarity on the
reactivity

CHAPTER 8.

Aromatic

Electrophilic

Substitution:

The arenium ion:

mechanism,

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Spectroscopy and
reactivity,
energy profile
diagrams; The
ortho/para
ratio, ipso
attack,
orientation in
other ring
systems;
Quantitative
treatment of
reactivity in
substrates and

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electrophiles;

Diazonium

coupling;

Vilsmeier

reaction;

Gattermann-Koch

reaction CHAPTER

9. Aromatic

Nucleophilic

Substitution:

The $ArSN_1$,

$ArSN_2$, Benzyne

and SRN_1

mechanisms;

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Reactivity – Spectroscopy

effect of
substrate
structure,
leaving group
and attacking
nucleophile; The
von Richter,
Sommelet-Hauser,
and Smiles
rearrangements

CHAPTER 10.

Elimination

Reactions: The

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E2, E1 and E1cB
mechanisms;
Orientation of
the double bond;
Reactivity
-effects of
substrate
structures,
attacking base,
the leaving
group and the
medium;
Mechanism and
orientation in

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pyrolytic

elimination

CHAPTER 11.

Addition to

Carbon-Carbon

Multiple Bonds:

Mechanistic and

stereochemical

aspects of

addition

reactions

involving

electrophiles,

nucleophiles and

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free radicals;
Regio- and chemos
electivity:
orientation and
reactivity;
Addition to
cyclopropane
ring;
Hydrogenation of
double and
triple bonds;
Hydrogenation of
aromatic rings;
Hydroboration;

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Michael

reaction;

Sharpless

asymmetric

epoxidation.

CHAPTER 12.

Addition to

Carbon-Hetero

Multiple Bonds:

Mechanism of

metal hydride

reduction of

saturated and

unsaturated

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carbonyl
compounds,
acids, esters
and nitriles;

Addition of
Grignard
reagents,
organozinc and
organolithium;

Reagents to
carbonyl and
unsaturated
carbonyl
compounds;

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Wittig reaction;
Mechanism of
condensation
reactions
involving
enolates -
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Knoevenagel,
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Mannich,
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reactions;
Hydrolysis of

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esters and
amides;
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esters.

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spectra of
molecules that
contain

transition-metal
ions. Chapter 2
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application of
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techniques to
the
photochemistry
of small
inorganic

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molecules, non-t
ransition-metal
compounds, and
transition-metal
complexes. The
remaining
chapters examine
several
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